## **AMENDMENTS TO THE CLAIMS**

This listing of claims will replace all prior versions and listings of claims in the application:

## **LISTING OF CLAIMS:**

1. (currently amended): A method for producing a complex metal oxide powder, comprising:

heating at least <a href="two">two</a> [[tow kinds of]] metal salts, or a complex metal salt comprising [[at least]] two <a href="different">different</a> [[kinds of]] metals, <a href="in-the-substantial absence of hydrogen halide gas">in the substantial absence of hydrogen halide gas</a>, to a temperature at which transition to a complex metal oxide occurs, and calcining the metal salts or the complex metal salt in the presence of a hydrogen halide gas.

- 2. (original): The method according to claim 1, wherein a concentration of the hydrogen halide gas after heating is from about 0.1 vol% to about 10 vol%.
- 3. (currently amended): The method according to claim 1,comprising <a href="heating at least three metal salts">heating at least three metal salts</a>, wherein at least one metal salt is a metal halide salt, and at least one metal salt is a [["]]non metal halide salt[["]].
- 4. (currently amended): The method according to claim 1, wherein the complex metal salt comprises at least two <u>different</u>[[kinds]] of metal atoms, a halogen atom, and a non-halogen atom.
- 5. (currently amended): The method according to claim 3, wherein each non metal halide salt is made of the same metal as one of the metal halide salts[[the metal halide salt and the non metal halide salt are made of a same metal]].

2

Amendment Under 37 C.F.R. § 1.111 U.S. Appln. 09/976,010 ATTORNEY DOCKET NO. Q66667

- 6. (original): The method according to claim 5, wherein the metal halide salt and the non metal halide salt made of the same metal are at a molar ratio in a range of 99.9:0.1 to 90:10.
- 7. (original): The method according to claim 1, wherein a temperature of the calcination is from about 500°C to about 1000°C.
- 8. (original): The method according to claim 1, wherein the complex metal oxide powder is a perovskite-structure oxide powder.
- 9. (original): The method according to claim 8, wherein the perovskite-structure oxide powder comprises a titanate.

3